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View Answer. Chemical equilibrium is

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established when the forward rate of reaction is equal to the reverse rate of reaction. a. TRUE b. FALSE. View Answer. For a reaction  $N_2 + O_2 \rightarrow 2NO$ . If  $K_c...$

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Chemical equilibrium (definition) The state reached when the concentration of

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reactants and products remain constant over time What is the general, reversible reaction for equilibrium?

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Answer and Explanation: A chemical reaction is a dynamic equilibrium. In chemical reactions when the system is at equilibrium, reactions are still occurring. However, the rate of forward reaction...

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**A chemical equilibrium is a dynamic equilibrium. True ...**

Section 17.2 Factors Affecting Chemical Equilibrium. 1. If a stress is applied to a system at equilibrium, the system shifts in the direction that relieves the stress.

2. Answers may vary, but should include changes in concentration, volume, and

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temperature. Some students may list pressure in place of volume. 3. left; remove N<sub>2</sub> or H<sub>2</sub>

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The equilibrium is represented by the

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equation above. (a) Write the expression for the solubility-product constant,  $K_{sp}$ , and calculate its value at  $18^{\circ}\text{C}$ . (b) Calculate the equilibrium concentration of  $\text{Mg}^{2+}$  in 1.000 liter of saturated  $\text{MgF}_2$  solution at  $18^{\circ}\text{C}$  to which 0.100 mole of solid  $\text{KF}$  has been added. The  $\text{KF}$  dissolves completely.

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What is chemical equilibrium? A state of balance in a reaction where the forward and reverse reaction speed is unequal. A state where the reaction is unbalanced.

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