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Using the electronic balance, measure and record (in Data Table C) the mass of a dry 50 mL beaker, flask, and graduated cylinder. Add 25.0 mL of distilled water to each piece of glassware. Add the last few drops carefully so that the bottom of the meniscus is exactly at the volume you want.

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Test tube B, which was a combination of distilled water and cobalt chloride, was an endothermic reaction. It was an endothermic reaction because when the test tube was heated, the solutions turned from pink to dark blue. This indicated that the equilibrium shifted to favor the

Restoring Balance: Equilibrium Lab by Cassidy Prater

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